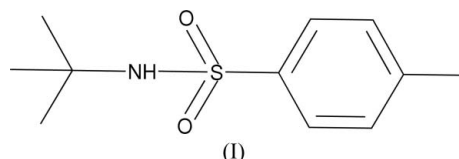
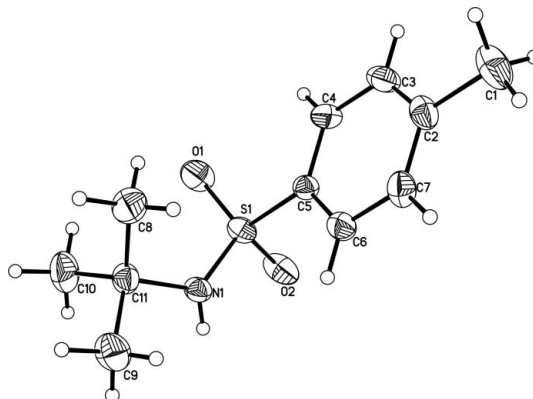


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yanyongping6@126.com**Key indicators**Single-crystal X-ray study
 $T = 291\text{ K}$
Mean $\sigma(\text{C}-\text{C}) = 0.003\text{ \AA}$
 R factor = 0.039
 wR factor = 0.109
Data-to-parameter ratio = 19.4For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.***N*-tert-Butyl-4-toluenesulfonamide**Molecules of the title compound, $\text{C}_{11}\text{H}_{17}\text{NO}_2\text{S}$, are linked by
paired $\text{N}-\text{H}\cdots\text{O}$ hydrogen bonds into centrosymmetric $R_2^2(8)$
dimers.Received 11 November 2006
Accepted 19 November 2006**Comment**The sulfonamide group is present in many bioactive
compounds (Patani & Lavoie, 1996; Thornber, 1979). As the
most widely used sulfonamide protecting group is the *tert*-
butyl group, *N*-*tert*-butylarylsulfonamides are usually used as
intermediates of sulfonamides (Le Bourdonnec *et al.*, 2000;
Murugesan *et al.*, 2003; Graham & Scholz, 1991). We report
here the crystal structure of the title compound, (I).Bond lengths and angles in (I) are normal. The dihedral
angle between the $\text{C}2-\text{C}7$ and $\text{S}1/\text{N}1/\text{C}11$ planes is $80.14(8)^\circ$.
The molecules form centrosymmetric dimers *via* $\text{N}-\text{H}\cdots\text{O}$
hydrogen bonds (Table 1 and Fig.2).**Experimental**Aqueous NaOH (7 ml, 10%) was added dropwise to a mixture of 4-
toluenesulfonyl chloride (0.58 g) and *tert*-butylamine (1 ml) with
constant stirring for 4 h. The solvent was then evaporated in a
vacuum. The residue was dissolved in aqueous ethanol and white
single crystals of the title compound suitable for X-ray diffraction
analysis were obtained by slow evaporation of the solvent.**Figure 1**
The molecular structure of (I), showing 30% probability displacement
ellipsoids and the atomic numbering.

Crystal data

$C_{11}H_{17}NO_2S$
 $M_r = 227.32$
 Monoclinic, $P2_1/c$
 $a = 6.1484$ (7) Å
 $b = 24.812$ (3) Å
 $c = 8.1623$ (9) Å
 $\beta = 97.306$ (1)°
 $V = 1235.1$ (2) Å³

$Z = 4$
 $D_x = 1.222$ Mg m⁻³
 Mo $K\alpha$ radiation
 $\mu = 0.24$ mm⁻¹
 $T = 291$ (2) K
 Block, white
 $0.47 \times 0.37 \times 0.25$ mm

Data collection

Bruker SMART CCD area-detector diffractometer
 φ and ω scans
 Absorption correction: multi-scan (SADABS; Sheldrick, 1996)
 $T_{min} = 0.895$, $T_{max} = 0.942$

8662 measured reflections
 2791 independent reflections
 2401 reflections with $I > 2\sigma(I)$
 $R_{int} = 0.016$
 $\theta_{max} = 27.5^\circ$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.039$
 $wR(F^2) = 0.109$
 $S = 1.08$
 2791 reflections
 144 parameters
 H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.0519P)^2 + 0.3046P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{max} = 0.013$
 $\Delta\rho_{max} = 0.27$ e Å⁻³
 $\Delta\rho_{min} = -0.31$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$N1-H1\cdots O2^i$	0.79 (2)	2.20 (2)	2.982 (2)	170 (2)

Symmetry code: (i) $-x, -y + 1, -z + 1$.

Atom H1 was located in a difference map and refined isotropically. The remaining H atoms were placed in calculated positions, and refined using a riding model, with $C-H = 0.93-0.96$ Å and $U_{iso}(H) = 1.2U_{eq}(C)$ or $1.5U_{eq}(\text{methyl } C)$.

Data collection: SMART (Bruker, 1998); cell refinement: SAINT (Bruker, 1998); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 1998); software used to prepare material for publication: SHELXTL.

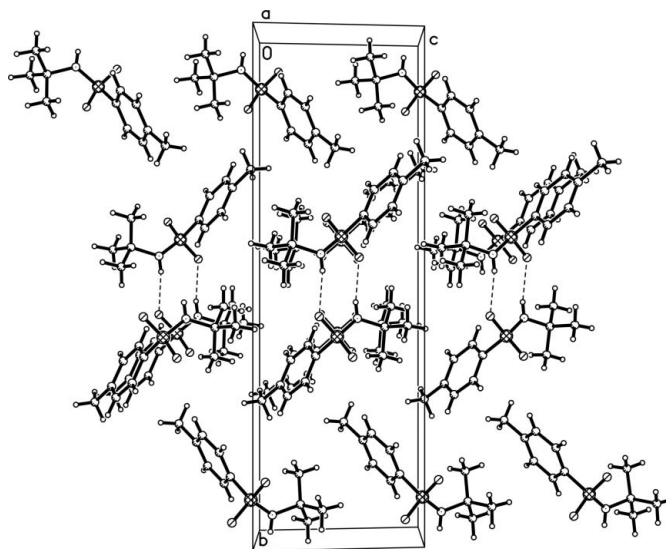


Figure 2

View of the packing along the a axis. $N-H\cdots O$ hydrogen bonds are shown as dashed lines.

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